

Supporting information for

Substantial Lifetime Enhancement for Si-Based Photoanodes Enabled by Amorphous TiO₂ Coating with Improved Stoichiometry

Yutao Dong¹, Mehrdad Abbasi², Jun Meng¹, Lazarus German¹, Corey Carlos¹, Jun Li¹, Ziyi
Zhang¹, Dane Morgan¹, Jinwoo Hwang², Xudong Wang^{1*}

¹Department of Materials Science and Engineering, University of Wisconsin-Madison, Madison,
WI, 53706, USA

²Department of Materials Science and Engineering, The Ohio State University, Columbus, OH
43210, USA

* Email: xudong.wang@wisc.edu

Supplementary Figures

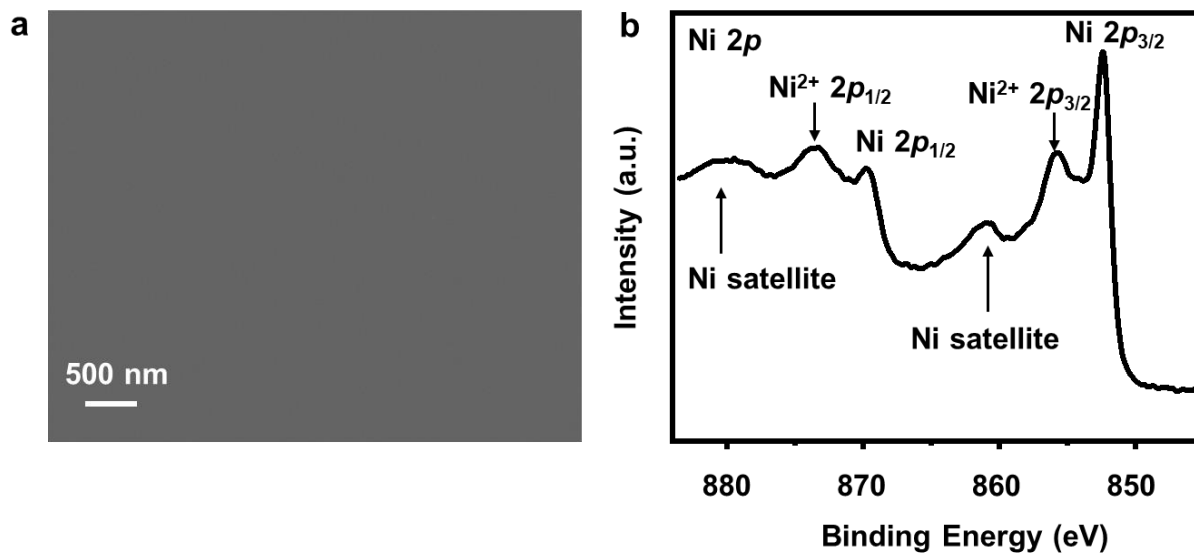


Figure S1. **a** Top-view SEM image of Ni sputtered Si/TiO₂ photoanode. **b** XPS Ni core spectrum of as-prepared Si/TiO₂/Ni photoanode.

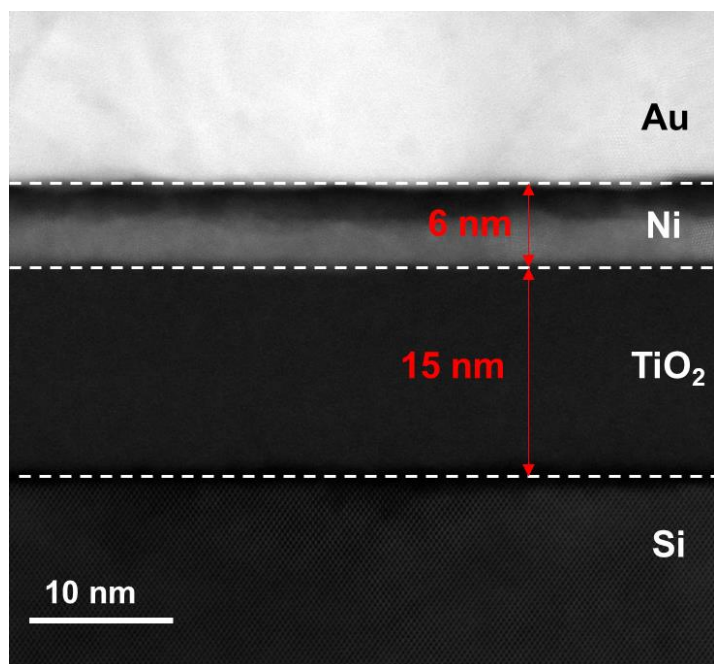


Figure S2. Cross-sectional HR-STEM images of as-prepared ~15 nm Si/TiO₂ with ~6 nm Ni sputtered.

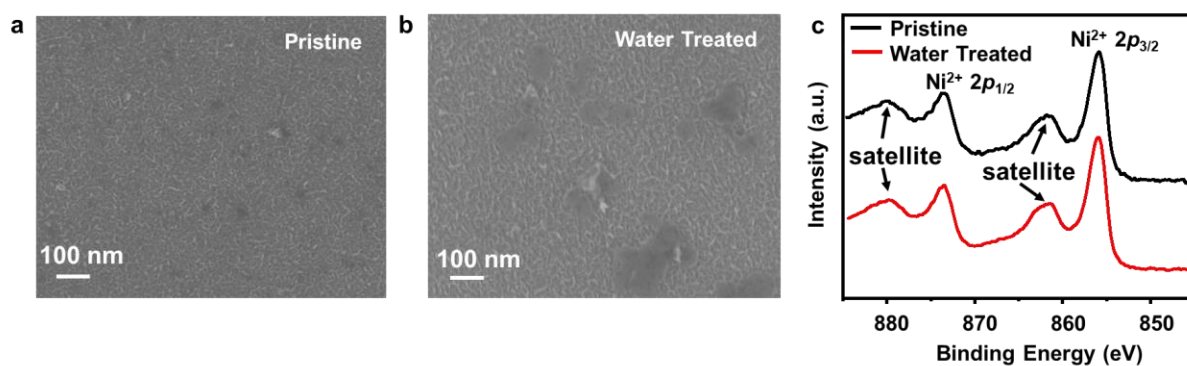


Figure S3. **a, b** SEM of Si/TiO₂/Ni after 1h test with pristine **(a)** and water treated **(b)** TiO₂ protection layer. **c** XPS Ni core spectra after 1h test of Si/TiO₂/Ni photoanode with pristine and water treated TiO₂ protection film.

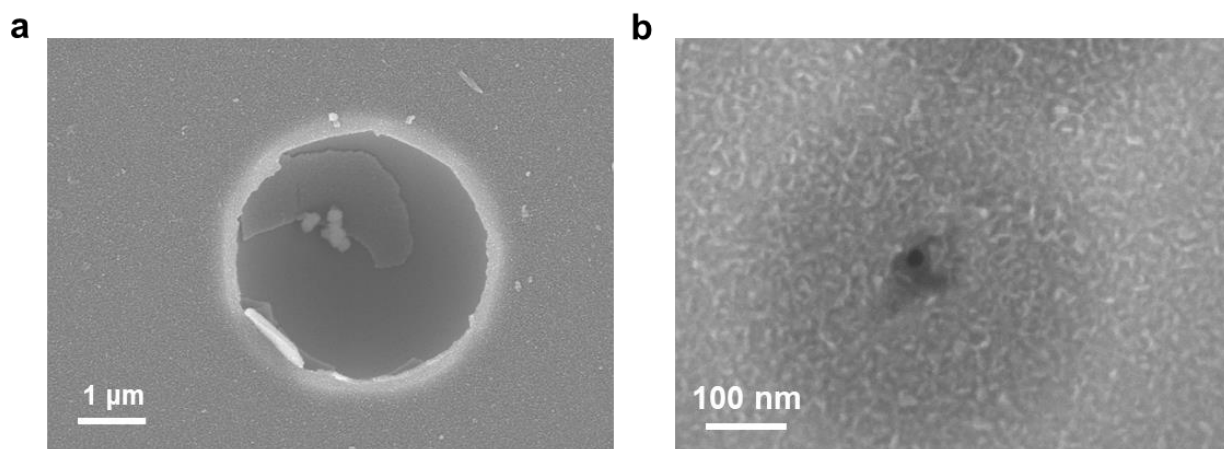


Figure S4. High magnification top-view SEM of pinholes in Si/TiO₂/Ni with pristine TiO₂ thin film during device failure. **(a)** 30h, **(b)** 5 h.

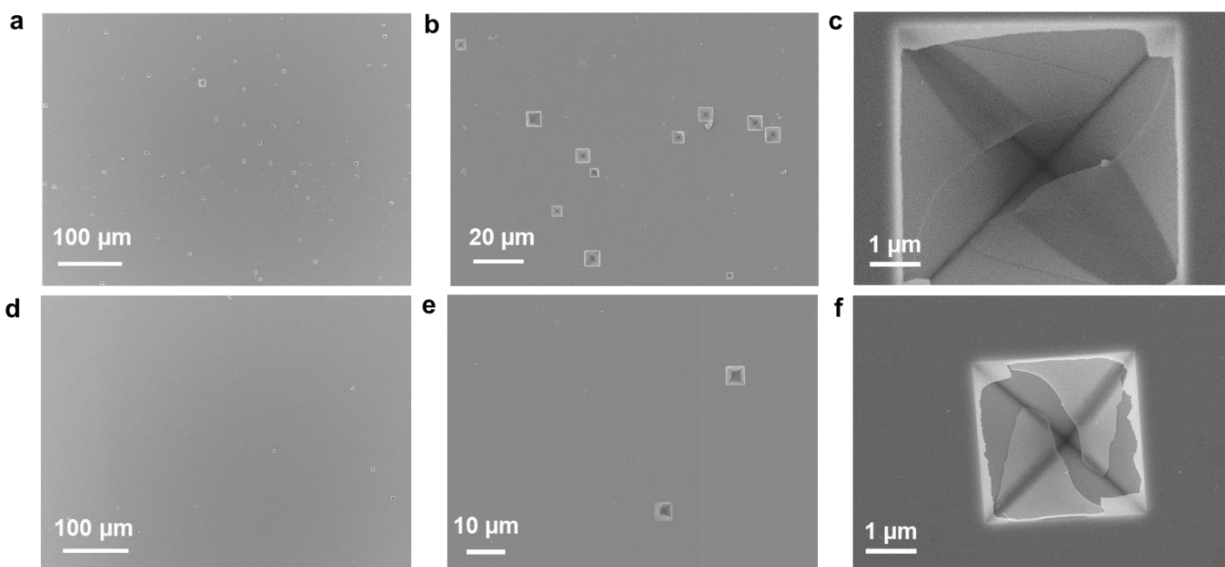


Figure S5. Chemical stability test of Si/TiO₂ of 1-day immersion in 1M KOH aqueous solution. Surface morphology of pristine (a)-(c), and water treated Si/TiO₂ (d)-(f) under different magnification.

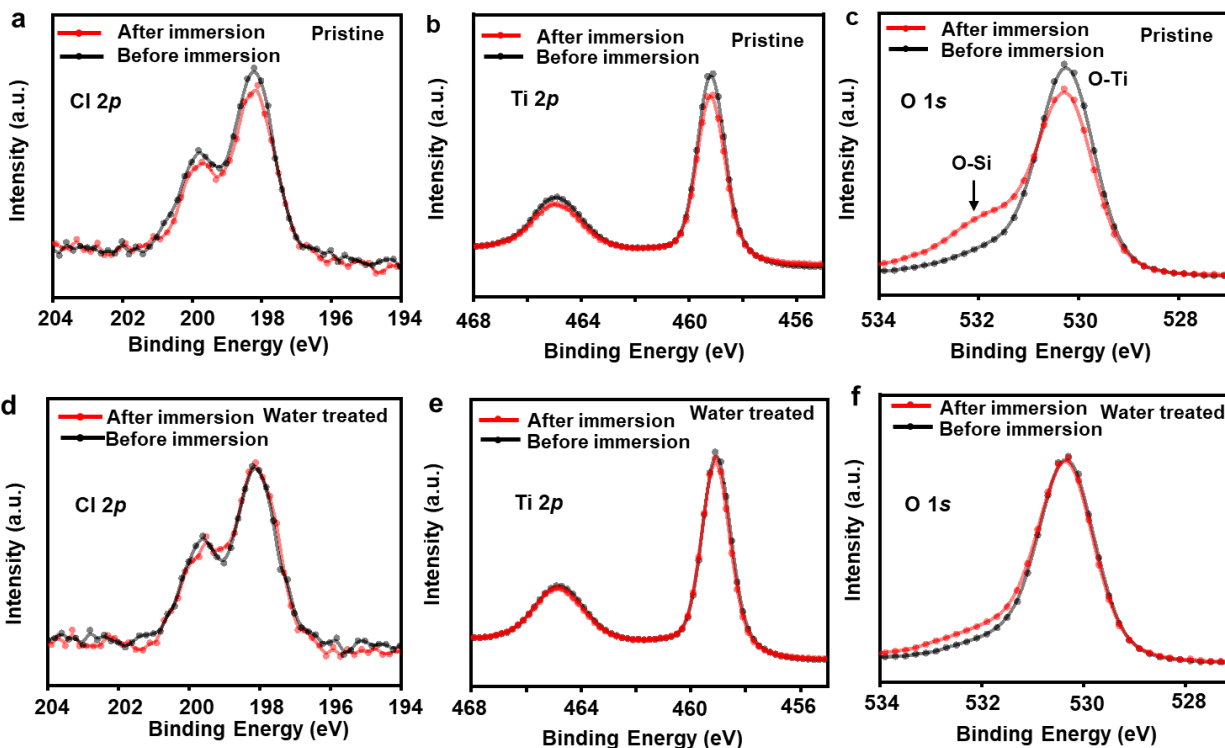


Figure S6. XPS core spectrum analysis of pristine, water treated Si/TiO₂ wafer before and after immersion in 1M KOH aqueous solution. (a) Cl 2*p*, (b) Ti 2*p*, (c) O 1*s* of Pristine TiO₂. (d) Cl 2*p*, (e) Ti 2*p*, (f) O 1*s* of water treated TiO₂.

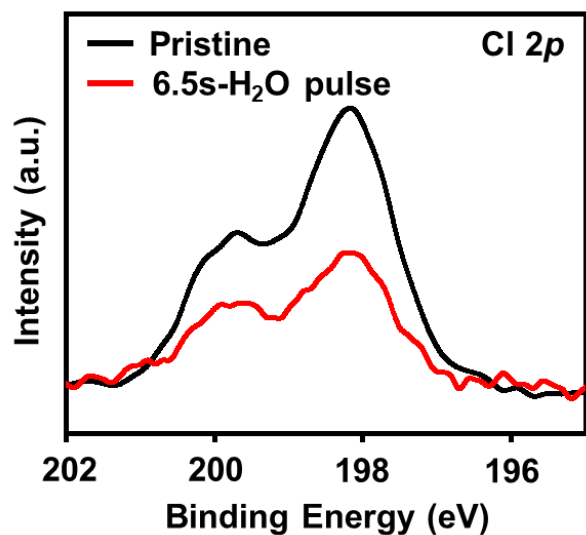


Figure S7. Cl 2p XPS core spectra of pristine TiO₂ (H₂O pulse was 0.5 s, black), and TiO₂ deposited using extended H₂O pulse (6.5 s, red).

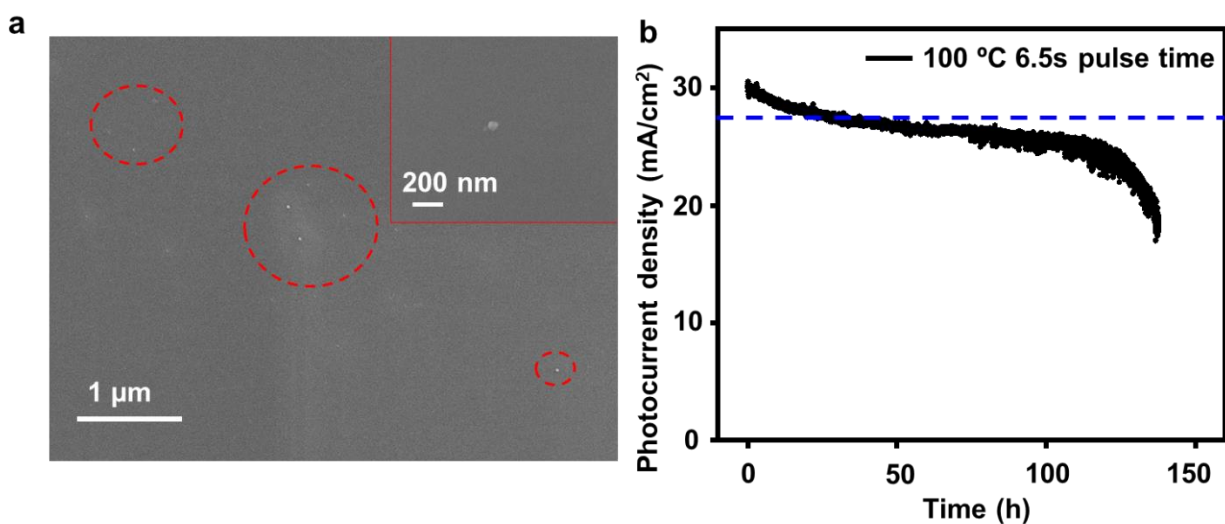


Figure S8. **a** Top-view SEM of 100 °C 200cycles Si/TiO₂ grown under elongated water pulse time of 6.5 s during deposition. **b** Chronoamperometry test of Si/TiO₂/Ni under the external bias of 1.8 V vs. RHE at 1 sun illumination. Blue dashed line marks the 90% threshold value.

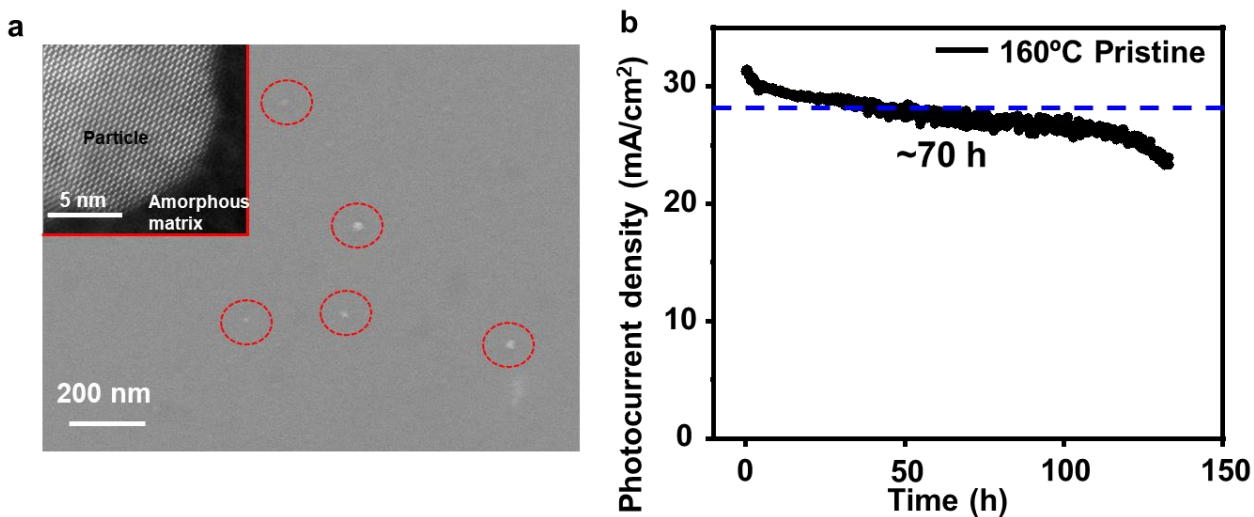


Figure S9. **a** Top-view SEM image of free-standing 160 °C-grown pristine TiO₂ films. Nanoparticles can be observed from the film surface, marked by red dashed circles. Inset is the HR-STEM image of the nanoparticle showing its crystalline phase. **b** Chronoamperometry test of Si/TiO₂/Ni under the external bias of 1.8V vs. RHE at 1 sun illumination. Blue dashed line marks the 90% threshold value.

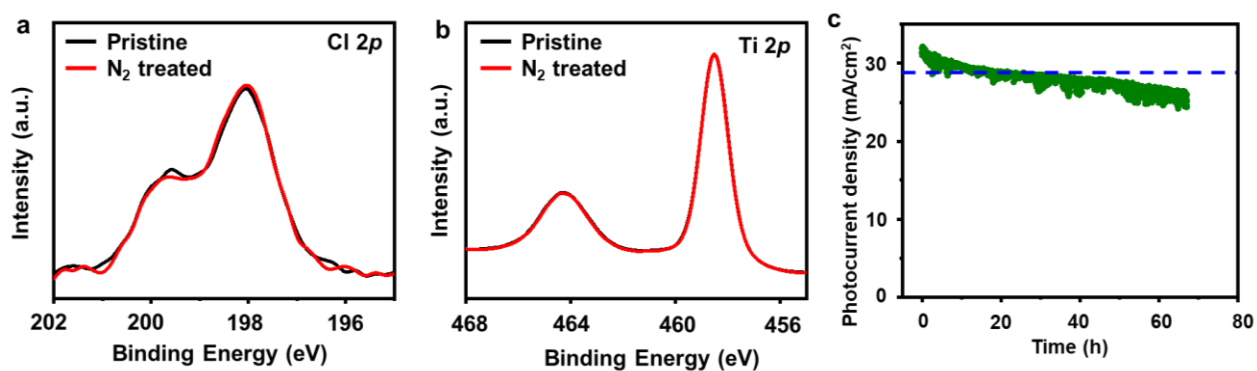


Figure S10. **a** Cl 2*p*, **b** Ti 2*p* XPS core spectra analysis of pristine and N₂ treated TiO₂ protection film. **c**, Chronoamperometry test of Si/TiO₂/Ni with 100 °C N₂ treated TiO₂ under the external bias of 1.8V vs. RHE at 1 sun illumination. Blue dashed line marks the 90% threshold value.

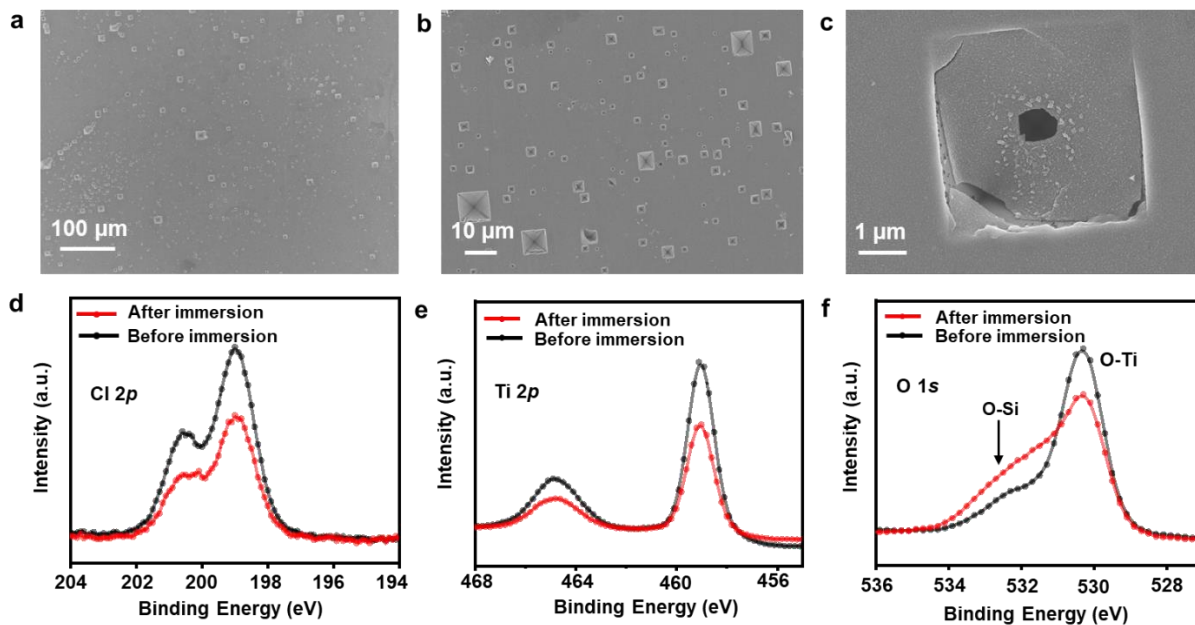


Figure S11. a-c Surface morphology of TiCl_4 treated Si/TiO_2 after 1 day immersion in 1M KOH. d-f Corresponding XPS core spectrums of TiCl_4 treated TiO_2 before and after immersion (d) Cl 2p, (e) Ti 2p, (f) O 1s.

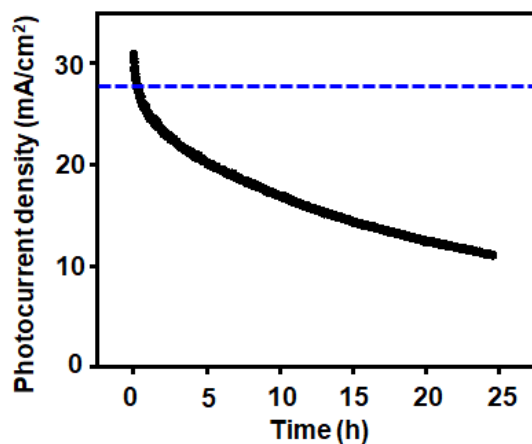


Figure S12. Chronoamperometry test of $\text{Si/TiO}_2/\text{Ni}$ with 100 $^\circ\text{C}$ TiCl_4 treated TiO_2 under the external bias of 1.8V vs. RHE at 1 sun illumination. Blue dashed line marks the 90% threshold value.

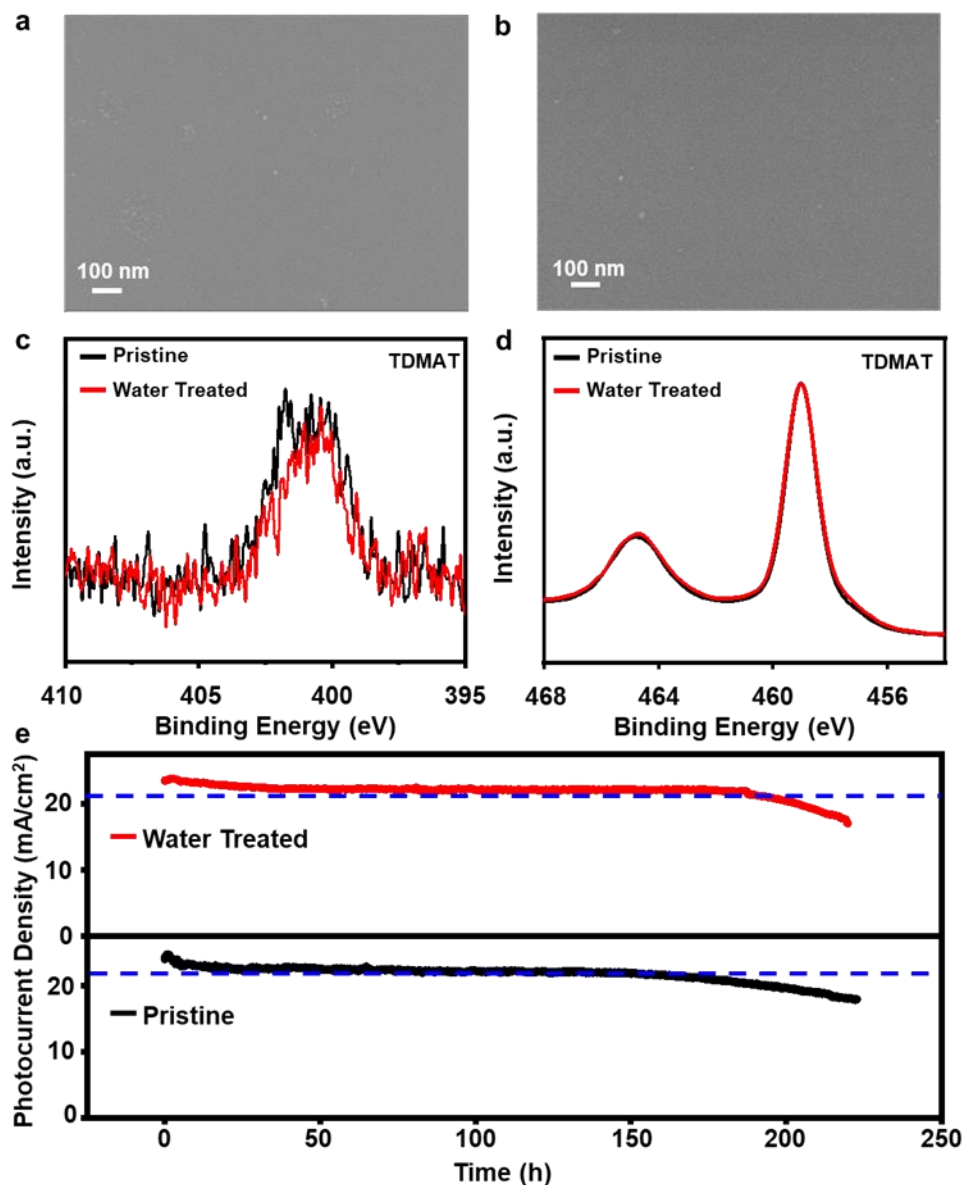


Figure S13. Si photoanode protected by pristine and water treated TiO₂ film from TDMAT precursor. **a, b** Top view SEM images of pristine and water treated TDMAT-TiO₂ film on n-Si substrate. **c** XPS N 1s peak of pristine and water treated TiO₂ film from TDMAT. **d** XPS Ti 2p peak of pristine and water treated TiO₂ film from TDMAT. **e** Chronoamperometry test of pristine and water treated TDMAT-TiO₂ protected Si photoanode measured in 1.0 M KOH aqueous solution under 1 sun illumination at an external bias of 1.8 V vs. RHE. Blue dashed line marks the 90% threshold value.